

1 In the Gibbs–Duhem relation $0 = SdT - VdP + Nd\mu$, we consider changes for constant T and N , that is $dT \rightarrow 0$, $dP \rightarrow (dP)_{T,N}$, and $d\mu \rightarrow (d\mu)_{T,N}$. Then

$$0 = 0 - V \left(\frac{\partial P}{\partial V} \right)_{T,N} + N \left(\frac{\partial \mu}{\partial V} \right)_{T,N} \quad \text{or} \quad N \left(\frac{\partial \mu}{\partial V} \right)_{T,N} = V \left(\frac{\partial P}{\partial V} \right)_{T,N}.$$

See also (2.9.14)–(2.9.16) in the lecture notes.

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(a) Below the critical temperature, there are molar volumes v for which

$$\frac{\partial P(T, v)}{\partial v} = P(T, v) \left[-\frac{1}{v-b} + \frac{a}{v^2 RT} \right] = 0$$

or

$$\left(v - \frac{a}{2RT} \right)^2 = \left(\frac{a}{2RT} \right)^2 - \frac{ab}{RT}.$$

This requires $RT < \frac{a}{4b}$, so that the critical temperature is $T_{\text{cr}} = \frac{a}{4bR}$ and the critical molar volume is $v_{\text{cr}} = \frac{a}{2RT_{\text{cr}}} = 2b$. For the critical pressure, we find

$$P_{\text{cr}} = P(T_{\text{cr}}, v_{\text{cr}}) = \frac{a}{4b^2} e^{-2}. \quad \text{Together, they give } \frac{P_{\text{cr}} v_{\text{cr}}}{T_{\text{cr}}} = 2e^{-2} R.$$

(b) We know that $\left. \frac{dP(T)}{dT} \right|_{T_{\text{cr}}} = \left. \frac{\partial P(T, v)}{\partial T} \right|_{T_{\text{cr}}, v_{\text{cr}}}$. Here, this gives

$$\left. \frac{dP(T)}{dT} \right|_{T_{\text{cr}}} = \left(\frac{1}{T} + \frac{a}{vRT^2} \right) P(T, v) \Big|_{T_{\text{cr}}, v_{\text{cr}}} = \left(1 + \frac{a}{v_{\text{cr}} RT_{\text{cr}}} \right) \frac{P_{\text{cr}}}{T_{\text{cr}}} = 3 \frac{P_{\text{cr}}}{T_{\text{cr}}}.$$

Therefore, we have

$$P(T) = \left(3 \frac{T}{T_{\text{cr}}} - 2 \right) P_{\text{cr}}$$

for temperatures just below the critical temperature.

3

(a) We have, quite simply,

$$\begin{aligned} \sum_{N=0}^{\infty} z^N Q(\beta, V, N) &= \sum_k e^{-\beta E_k} \sum_{N=0}^{\infty} z^N \delta_{N, N_k} = \sum_k e^{-\beta E_k} z^{N_k} \\ &= \sum_k e^{-\beta E_k} e^{\beta \mu N_k} = \sum_k e^{-\beta(E_k - \mu N_k)} = Z(\beta, V, z). \end{aligned}$$

(b) Since

$$Z(\beta, V, z) = e^{Vz/\lambda^3} = \sum_{N=0}^{\infty} \frac{1}{N!} \left(\frac{Vz}{\lambda^3} \right)^N = \sum_{N=0}^{\infty} \frac{z^N}{N!} \left(\frac{V}{\lambda^3} \right)^N,$$

we read off that $Q(\beta, V, N) = \frac{1}{N!} \left(\frac{V}{\lambda^3} \right)^N$.

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(a) F is intensive because it has the same value independent of the number of particles.

(b) The canonical partition function is

$$Q(\beta, F, N) = \frac{1}{N!} \left[\int \frac{(d\mathbf{r}) (d\mathbf{p})}{(2\pi\hbar)^3} e^{-\beta(\frac{1}{2m}\mathbf{p}^2 + Fr)} \right]^N,$$

where [see Problem 3(b)]

$$\int \frac{(d\mathbf{p})}{(2\pi\hbar)^3} e^{-\beta\frac{1}{2m}\mathbf{p}^2} = \frac{1}{\lambda^3}$$

and

$$\int (d\mathbf{r}) e^{-\beta Fr} = 4\pi \int_0^{\infty} dr r^2 e^{-\beta Fr} = \frac{8\pi}{(\beta F)^3}.$$

Accordingly, we have

$$Q(\beta, F, N) = \frac{1}{N!} \left(\frac{8\pi}{(\lambda\beta F)^3} \right)^N.$$

(c) We have, $\frac{1}{N} \langle E \rangle = -\frac{1}{N} \frac{\partial}{\partial \beta} \log Q(\beta, F, N) = -\frac{9}{2\beta} = \frac{9}{2} k_B T$ since $Q \propto \beta^{-9N/2}$.

(d) We note that the kinetic energy is inversely proportional to the mass m and the potential energy is proportional to F , and $Q(\beta, F, N) \propto m^{3N/2} F^{-3N}$. Therefore, we have

$$\frac{1}{N} \langle E_{\text{kin}} \rangle = \frac{m}{N\beta} \frac{\partial}{\partial m} \log Q(\beta, F, N) = \frac{3}{2\beta} = \frac{3}{2} k_B T = \frac{1}{3} \langle E \rangle$$

and

$$\frac{1}{N} \langle E_{\text{pot}} \rangle = -\frac{F}{N\beta} \frac{\partial}{\partial F} \log Q(\beta, F, N) = \frac{3}{\beta} = 3k_B T = \frac{2}{3} \langle E \rangle.$$

Clearly, $\langle E_{\text{kin}} \rangle + \langle E_{\text{pot}} \rangle = \langle E \rangle$.
