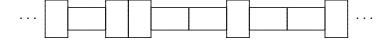
1. Warming up (10 marks)

You add a milli-calorie of heat ($\simeq 4 \times 10^{-3} \, \mathrm{J}$) to a certain substance at a temperature of $300 \, \mathrm{K}$ (water, perhaps, but that is not relevant). What is the corresponding change in the number of microstates available to the system?

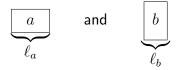
Hint: Just get an estimate by a back-of-the-envelope calculation.

2. A simple rubber-band model (30=8+6+8+8 marks)

A physical model of a rubber band is a single long chain in which N molecules are linked:



Each molecule has two possible configurations:



which have lengths ℓ_a, ℓ_b and energies $E_a = 0$ and $E_b = E_0 > 0$. The rubber band is in thermal contact with the atmosphere at temperature $T = 1/(k_{\rm B}\beta)$.

- (a) Determine the canonical partition function $Q(\beta, N)$ for this system and then find the entropy $S(\beta, N)$.
- **(b)** What is the average number of molecules with energy E_a ?
- (c) What is $\Omega(E, N)$, the number of microstates with energy E? Is the entropy obtained as $S = k_{\rm B} \log \Omega$ equal to the entropy found in part (a)?
- (d) Find the average length L(T,N) of the rubber band. When the temperature is increased, does L(T,N) increase or decrease? What is L(T,N) for very low temperatures $(k_{\rm B}T\ll E_0)$ and for very high temperatures $(k_{\rm B}T\gg E_0)$?

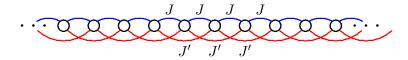
3. A dilute gas (30=5+7+14+4 marks)

The grand-canonical partition function $Z(\beta,V,z)$ of a dilute gas is given by $\log Z=(k_{\rm\scriptscriptstyle B}T_0\beta)^{-\kappa}\frac{V}{V_0}z$, where T_0 , V_0 , and κ are positive material constants.

- (a) Confirm that PV = constant for isothermal changes.
- (b) What is the equation of state for isentropic changes? Confirm that you get the expected answer when $\kappa=\frac{3}{2}.$
- (c) Determine the free energy F(T, V, N) and the internal energy U(S, V, N).
- (d) What are the heat capacities for constant volume and constant pressure?

4. Ising with next-next-neighbor interaction (30=4+4+12+10 marks)

Consider a modified Ising chain (or ring) with N sites, no on-site energy, and next-neighbor interaction strength J. There is also a next-next-neighbor interaction of strength J'. Symbolically, then, we have this picture:



As usual, we use $K = \beta J$ and $K' = \beta J'$ for convenience.

- (a) What is the free energy F(K, 0, N) when K' = 0?
- **(b)** What is the free energy $F(0,K^{\prime},N)$ when K=0?
- (c) For $K \neq 0$ and $K' \neq 0$, find the canonical partition function and then find the free energy F(K, K', N). Verify that you get the expected expressions when K' = 0 or K = 0.
- (d) Determine the heat capacity to the leading order in K' when $0 < K' \ll 1$.

Hint: Remember that $s_{j-1}s_{j+1} = (s_{j-1}s_j)(s_js_{j+1}).$

End of Paper

(BG Englert)