

PC2135 Thermodynamics and Statical Mechanics, homework 6 (last set)

Due Tuesday 18 Apr 11:59PM

Problems 7.8 (28 marks (a) to (d), 7 marks each). Problem 7.26 (24 marks, each subquestion is 8 marks for the rest). Problem 7.43 (24 marks). 7.44 (24 marks). Total 100 marks.

**Problem 7.8.** Suppose you have a “box” in which each particle may occupy any of 10 single-particle states. For simplicity, assume that each of these states has energy zero.

- (a) What is the partition function of this system if the box contains only one particle?
- (b) What is the partition function of this system if the box contains two distinguishable particles?
- (c) What is the partition function if the box contains two identical bosons?
- (d) What is the partition function if the box contains two identical fermions?

**Problem 7.26.** In this problem you will model helium-3 as a noninteracting Fermi gas. Although  ${}^3\text{He}$  liquefies at low temperatures, the liquid has an unusually low density and behaves in many ways like a gas because the forces between the atoms are so weak. Helium-3 atoms are spin-1/2 fermions, because of the unpaired neutron in the nucleus.

- (a) Pretending that liquid  ${}^3\text{He}$  is a noninteracting Fermi gas, calculate the Fermi energy and the Fermi temperature. The molar volume (at low pressures) is  $37 \text{ cm}^3$ .
- (b) Calculate the heat capacity for  $T \ll T_F$ , and compare to the experimental result  $C_V = (2.8 \text{ K}^{-1})NkT$  (in the low-temperature limit). (Don't expect perfect agreement.)
- (c) The entropy of *solid*  ${}^3\text{He}$  below 1 K is almost entirely due to its multiplicity of nuclear spin alignments. Sketch a graph  $S$  vs.  $T$  for liquid and solid  ${}^3\text{He}$  at low temperature, and estimate the temperature at which the liquid and solid have the same entropy. Discuss the shape of the solid-liquid phase boundary shown in Figure 5.13.

**Problem 7.43.** At the surface of the sun, the temperature is approximately 5800 K.

- (a) How much energy is contained in the electromagnetic radiation filling a cubic meter of space at the sun's surface?
- (b) Sketch the spectrum of this radiation as a function of photon energy. Mark the region of the spectrum that corresponds to visible wavelengths, between 400 nm and 700 nm.
- (c) What fraction of the energy is in the visible portion of the spectrum? (Hint: Do the integral numerically.)

**Problem 7.44.** Number of photons in a photon gas.

- (a) Show that the number of photons in equilibrium in a box of volume  $V$  at temperature  $T$  is

$$N = 8\pi V \left(\frac{kT}{hc}\right)^3 \int_0^\infty \frac{x^2}{e^x - 1} dx.$$

The integral cannot be done analytically; either look it up in a table or evaluate it numerically.

- (b) How does this result compare to the formula derived in the text for the entropy of a photon gas? (What is the entropy per photon, in terms of  $k$ ?)
- (c) Calculate the number of photons per cubic meter at the following temperatures: 300 K; 1500 K (a typical kiln); 2.73 K (the cosmic background radiation).