

Suggested Solution for Statistical Physics and Thermodynamics AY2007-08 Semester 2

NUS Physics Society

1. (a) (Note that the subscript 1, 2 indicating the subsystem 1 and subsystem 2 respectively. Whereas the variables without subscript refer to variables of whole system)

$$\begin{aligned} S_1 + S_2 &= S & V_1 + V_2 &= V \\ E_1 + E_2 &= E & N_1 + N_2 &= N \end{aligned} \quad (1)$$

Hence, $\frac{\partial V_2}{\partial V_1} = -1$ as $\frac{\partial V}{\partial V_1} = 0$

$$\left(\frac{\partial S_1}{\partial V_1}\right)_{E_1, N_1} = \left(\frac{\partial S_2}{\partial V_2}\right)_{E_2, N_2} \quad (2)$$

The pressure of each subsystem is defined as $P_i = T_i \frac{\partial S_i}{\partial V_i}$ where $i = 1$ or 2 .

Hence the equilibrium condition is achieved when $T_1 = T_2$ and $P_1 = P_2$ from equation ??.

(b)

$$\frac{dS}{dt} = \frac{\partial S_1}{\partial V_1} \frac{dV_1}{dt} + \frac{\partial S_2}{\partial V_2} \frac{dV_2}{dt} = \left(\frac{P_1}{T_1} - \frac{P_2}{T_2}\right) \frac{dV_1}{dt} > 0 \quad \text{from the second law of thermodynamics.}$$

Hence, if $P_1 > P_2$, $\frac{dV_1}{dt} > 0$ according to the second law.

2. (a)

$$\begin{aligned} dS &= \frac{dQ}{T} & dQ &= C \cdot dT \\ \Delta S &= C \int_{T_1}^{T_0} \frac{1}{T} dT = C \ln\left(\frac{T_0}{T_1}\right) \end{aligned}$$

- (b) Change in entropy of water = $mC \cdot \ln\left(\frac{T_0}{T_1}\right) = 1000 \times 4.2 \times \ln(373/273) = 1311 JK^{-1}$
 Change in entropy of heat reservoir = $-1311 JK^{-1}$
 while change in entropy of the entire system = 0

3. (a)

$$\begin{aligned} S &= - \left(\frac{\partial F}{\partial T}\right)_V \\ &= NK \ln \left[\frac{eV}{N} \left(\frac{2\pi mkT}{h^2}\right)^{3/2} \right] + \frac{3NKT^{3/2}}{2} \end{aligned}$$

(b)

4. (a) Vapour is much less dense than water, $\Delta V = V_{vapour} - V_{liquid} \approx V_{vapour}$

$$\begin{aligned} PV &= RT \quad \text{per one mole} \\ \text{Hence, } L &= T \Delta V \frac{dP}{dT} \\ &= \frac{RT^2}{P} \frac{dP}{dT} \end{aligned}$$

(b)

$$\begin{aligned} \frac{dP}{dT} &= \frac{(788 - 733.7)\text{mmHg}}{(373 - 272)\text{K}} \\ &= \frac{54.3\text{mmHg}}{2\text{K}} = 7239\text{PaK}^{-1} \\ L &= \frac{RT^2}{P} \frac{dP}{dT} \\ &= \frac{8.31 \times 373^2}{760 \times 133.3} \times 7239 = 82614\text{J/mole} \end{aligned}$$

5. (a) Fermi energy, ϵ_F achieved at the condition $T = 0\text{K}$. The step function of the equation is ignored. Thus,

$$\begin{aligned} N(\epsilon_F) &= \frac{4\pi V}{h^3} (2m)^{3/2} \int_0^{\epsilon_F} \epsilon^{1/2} d\epsilon \\ &= \frac{4\pi V}{h^3} \frac{2}{3} \epsilon_F^{3/2} \end{aligned}$$

$$\text{Hence, after rearranging, } \epsilon_F = \frac{h^2}{2m} \left(\frac{3N}{8\pi V} \right)^{2/3}$$

- (b) $\epsilon_F = k \cdot T_F$ $\epsilon_F = \frac{1}{2} m V_F^2$ where k is Boltzmann constant.

- (c) Given $\frac{N}{V} = 4.7 \times 10^{22} \text{cm}^{-3} = 4.7 \times 10^{28} \text{m}^{-3}$.

$$\begin{aligned} \epsilon_F &= \frac{h^2}{2m} \left(\frac{3}{8\pi} \frac{N}{V} \right)^{2/3} \\ &= \frac{(6.626 \times 10^{-34})^2}{2 \times 9.11 \times 10^{-31}} \left(\frac{3}{8\pi} \times 4.7 \times 10^{28} \right)^{2/3} = 7.608 \times 10^{-19} \text{J} \end{aligned}$$

$$T_F = \frac{7.608 \times 10^{-19}}{1.381 \times 10^{-23}} = 5.509 \times 10^4 \text{K}$$

$$V_F = \sqrt{\frac{2\epsilon_F}{m}} = \sqrt{\frac{2 \times 7.608 \times 10^{-19}}{9.11 \times 10^{-31}}} = 1.292 \times 10^6 \text{m/s}$$

6. Given, $Q_P = kT^2 \frac{d}{dT} \ln K_p(T)$

$$\begin{aligned}
 Q_V &= Q_P - kT \sum_i v_i = kT^2 \frac{d}{dT} \ln K_p(T) - kT \sum_i v_i \\
 &= kT^2 \frac{d}{dT} \sum_i v_i \ln[kT f_i(T)] - kT \sum_i v_i \\
 &= kT^2 \frac{d}{dT} \sum_i v_i \ln[f_i(T)] + kT^2 \frac{d}{dT} \sum_i v_i \ln[kT] - kT \sum_i v_i \\
 &= kT^2 \frac{d}{dT} \sum_i v_i \ln[f_i(T)] = kT^2 \frac{d}{dT} \ln \left[\prod_i (f_i(T))^{v_i} \right] \\
 &= kT^2 \frac{d}{dT} \ln K_c(T)
 \end{aligned}$$

Long Questions

1. (a) The energy of the dipole with magnetic moment parallel to magnetic field, $= -B \cdot \mu$.
The energy of the dipole with magnetic moment anti-parallel to magnetic field, $= B \cdot \mu$.

$$\begin{aligned}
 \ln Z &= \sum_i \exp(-\beta \epsilon_i) = \exp(\beta B \mu) + \exp(-\beta B \mu) \\
 \frac{\partial \ln Z}{\partial B} &= \beta \mu \left(\frac{1}{Z} \right) \exp(\beta B \mu) + (-\beta \mu) \left(\frac{1}{Z} \right) \exp(-\beta B \mu) \\
 \frac{1}{\beta} \left(\frac{\partial \ln Z}{\partial B} \right)_\beta &= \mu \left(\frac{1}{Z} \right) \exp(\beta B \mu) + (-\mu) \left(\frac{1}{Z} \right) \exp(-\beta B \mu)
 \end{aligned}$$

We know that population of dipole with magnetic moment parallel (anti-parallel) to magnetic field $= \frac{1}{Z} \exp(\beta B \mu)$ ($\frac{1}{Z} \exp(-\beta B \mu)$). Hence,

$$\begin{aligned}
 \bar{\mu} &= P_{(+)} \mu + P_{(-)} (-\mu) = \frac{1}{\beta} \left(\frac{\partial \ln Z}{\partial B} \right)_\beta \\
 - \left(\frac{\partial \ln Z}{\partial \beta} \right)_B &= (-B \mu) \left(\frac{1}{Z} \right) \exp(\beta B \mu) + (B \mu) \left(\frac{1}{Z} \right) \exp(-\beta B \mu) \\
 &= P_{(+)} \epsilon_+ + P_{(-)} \epsilon_- = \bar{E}.
 \end{aligned}$$

(b)

$$\begin{aligned}
 I &= \frac{N \cdot \bar{\mu}}{V} = \frac{N}{V} \times \left[\mu \left(\frac{1}{Z} \right) \exp(\beta B \mu) + (-\mu) \left(\frac{1}{Z} \right) \exp(-\beta B \mu) \right] \\
 &= \frac{N}{V} \mu \times \frac{[\exp(x) - \exp(-x)]}{\exp(x) + \exp(-x)} \text{ where } x = \beta B \mu \\
 &= \frac{N}{V} \mu \tanh x
 \end{aligned}$$

In the limit of low magnetic field and high temperature, $x \ll 1$ and thus $\tanh x \approx x$. Thus,
 $I = \frac{N}{V} \mu x = \frac{N \mu^2 B}{V k T}$

Magnetic susceptibility $X = \frac{I}{H} = \frac{N\mu^2\mu_0}{VkT}$ where $H = \frac{B}{\mu_0}$
 $X \propto \frac{1}{T}$ Curie's law is verified.

(c) $\Omega(n) = \frac{N!}{n!(N-n)!}$ statistical weight of n dipoles parallel to magnetic field from total N dipoles.

$S(n) = k \ln[\Omega(n)] = k[N \ln(N) - n \ln(n) - (N-n) \ln(N-n)]$
 Stirling approximation $\ln(n!) \approx n \ln(n) - n$

$$\begin{aligned} \frac{1}{T} &= \left(\frac{\partial S}{\partial E} \right) = \frac{\partial S(n)}{\partial n} \cdot \frac{\partial n}{\partial E} \\ &= \left[k \ln\left(\frac{N-n}{n}\right) \right] \left[-\frac{1}{2\mu B} \right] \text{ as } E = (N-2n)\mu B \\ &= \frac{k}{2\mu B} \ln\left(\frac{n}{N-n}\right) \end{aligned}$$

Hence, $\beta = \frac{1}{2\mu B} \ln\left(\frac{n}{N-n}\right)$ and If $n < \frac{1}{2}N$, $T < 0$ (negative temperature)

2. (a) Given $P(v)dv = \frac{4}{\sqrt{\pi}}u^2 \exp(-u^2)du = P(u)du$

$$\begin{aligned} \frac{dP(u)}{du} &= \frac{d}{du} \left(\frac{4}{\sqrt{\pi}} u^2 \exp(-u^2) \right) \\ &= \frac{-8}{\sqrt{\pi}} u^3 \exp(-u^2) + \frac{8}{\sqrt{\pi}} u \cdot \exp(-u^2) \\ &= u \cdot \exp(-u^2) \frac{8}{\sqrt{\pi}} (1 - u^2) \end{aligned}$$

Hence, when $u = 1$, most probable speed achieved. Which imply that $V_{max} = \left(\frac{2kT}{m}\right)^{1/2}$.

(b)

$$\begin{aligned} \bar{v} &= \int_0^\infty v P(v) dv = \int_0^\infty u \cdot \left(\frac{2kT}{m}\right)^{1/2} \frac{4}{\sqrt{\pi}} u^2 \exp(-u^2) du \\ &= \left(\frac{2kT}{m}\right)^{1/2} \frac{4}{\sqrt{\pi}} \int_0^\infty u^3 \exp(-u^2) du \\ &= \left(\frac{2kT}{m}\right)^{1/2} \frac{4}{\sqrt{\pi}} \left(\frac{1}{2}\right) \quad \text{from identity} \\ &= 2 \left(\frac{2kT}{\pi m}\right)^{1/2} \end{aligned}$$

(c)

$$\begin{aligned} V_{rms}^2 &= \int_0^\infty v^2 P(v) dv = \int_0^\infty u^2 \cdot \left(\frac{2kT}{m}\right) \frac{4}{\sqrt{\pi}} u^2 \exp(-u^2) du \\ &= \left(\frac{2kT}{m}\right) \frac{4}{\sqrt{\pi}} \int_0^\infty u^4 \exp(-u^2) du \\ &= \left(\frac{2kT}{m}\right) \frac{4}{\sqrt{\pi}} \left(\frac{3}{8}\sqrt{\pi}\right) \quad \text{from identity} \\ &= \frac{3kT}{m} \end{aligned}$$

Hence, $V_{rms} = \left(\frac{3kT}{m}\right)^{1/2}$

(d)

$$\bar{E} = \frac{1}{2}m \cdot V_{rms}^2 = \frac{3}{2}kT$$

3. (a) Ordinary, the number of particles of BE gas, N is given as

$$V \frac{2\pi(2m)^{3/2}}{h^3} \int_0^\infty \frac{\epsilon^{1/2} d\epsilon}{e^{\beta(\epsilon-\mu)} - 1}$$

When, $T \downarrow$, μ must \uparrow for N/V to be kept constant. Meanwhile, $T = T_c$ imply $\mu = 0$. Hence,

$$\frac{N}{V} = \left(\frac{2\pi mkT_c}{h^2}\right)^{3/2} \left(\frac{2}{\sqrt{\pi}} \int_0^\infty \frac{z^{1/2} dz}{e^z - 1}\right) \quad \text{with} \quad z = \frac{\epsilon}{kT_c}$$

Thus, $T_c = \frac{1}{k} \left(\frac{N}{V}\right)^{2/3} \frac{h^2}{2\pi m} \left(\frac{1}{2.612}\right)^{2/3}$

(b) Below T_c , the chemical potential is extremely closed to zero. The number of particles with non-zero energy can be computed as

$$N_{\epsilon>0} = V \frac{2\pi(2m)^{3/2}}{h^3} \int_0^\infty \frac{\epsilon^{1/2} d\epsilon}{e^{\beta\epsilon} - 1}$$

Introduce $z = \beta\epsilon$, $N_{\epsilon>0} = V \left(\frac{2\pi mkT}{h^2}\right)^{3/2} \left(\frac{2}{\sqrt{\pi}} \int_0^\infty \frac{z^{1/2} dz}{e^z - 1}\right) = N \left(\frac{T}{T_c}\right)^{3/2}$

Meanwhile, for number of particles with zero-energy ground state is $N(1 - (T/T_c)^{3/2})$.

(c) Energy of BE gas at $T < T_c$, and introduce $z = \beta\epsilon$

$$\begin{aligned} E &= V \left(\frac{2\pi mkT}{h^2}\right)^{3/2} \left(\frac{2}{\sqrt{\pi}} \int_0^\infty \frac{z^{3/2} dz}{e^z - 1}\right) \\ N &= V \left(\frac{2\pi mkT_c}{h^2}\right)^{3/2} \left(\frac{2}{\sqrt{\pi}} \int_0^\infty \frac{z^{1/2} dz}{e^z - 1}\right) \end{aligned}$$

Hence, $E = 0.77Nk \frac{T^{5/2}}{T_c^{3/2}}$ computed with the identity given by question. Heat capacity, $C_v = \frac{dE}{dT} = \frac{5}{2} \times 0.77NR \left(\frac{T}{T_c}\right)^{3/2} = 1.93NR \left(\frac{T}{T_c}\right)^{3/2}$

(d) Bose-Einstein condensation is different from ordinary vapour-liquid condensation in the way that no *spatial* separation into phases in BE gas.