Amide/Hydride system for hydrogen storage



Li-N-H system — A milestone in the history of hydrogen storage materials



Xiong Z T, Chen P, Wu G T, et al., J. Mater. Chem. 2003, 13, 1676

Li-Mg-N-H system — One of the most promising solid materials for on-board hydrogen storage

attraction



 $2\text{LiNH}_2 + \text{MgH}_2 \rightarrow \text{Li}_2\text{Mg(NH)}_2 + 2\text{H}_2 \leftrightarrow \text{Mg(NH}_2)_2 + 2\text{LiH}$ $\Delta \text{H} = -38 \text{ kJ/mol-H}_2$

5.88wt%

Z. T. Xiong, G. T. Wu, J. J. Hu and P. Chen, Adv. Mater. 2004, 16, 1522

<u>Li-Ca-N-H system — Applying amide-hydride interaction into material synthesis</u>

Imides used to be prepared from the thermal decomposition of corresponding amides. However, the lack of amide materials always leads to the unavailability of imides, especially in the case of ternary or higher imides. By dehydrogenation of lithium amide and calcium hydride mixture at given temperature new ternary imides, $Li_2Ca(NH)_2$, were synthesized in our group and preliminary investigations over its unique structures have been carried out.



X-ray diffraction patterns of the ternary imide $Li_2Ca(NH)_2$. + observed; — calculated; bottom line - the difference; and vertical tick - Bragg peak positions







K-edge of the Li,Ca(NH),

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